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Enantioselective Synthesis of Bicyclic δ -Lactones via N-Heterocyclic Carbene-Catalyzed Cascade Reaction

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S Supporting Information

[AB](#page-2-0)STRACT: [The N-hetero](#page-2-0)cyclic carbene-catalyzed cascade reaction of enals with malonates to give bicyclic δ -lactones was developed. The cyclopentane- and cyclohexane-fused δlactones with three continued stereocenters were obtained in high yields with excellent diastereo- and high enantioselectivities.

In the past, N-heterocyclic carbenes (NHCs) have been
demonstrated as powerful catalysts for various organic
reactions. Beyond the classical NHC catalyzed benzoin demonstrated as powerful catalysts for various organic reactions. Beyond the classical NHC-catalyzed benzoin reactions¹ and Stetter reactions² of aldehydes, the NHCcatalyzed extended umpolung of functionalized aldehydes, such as enals^{[3](#page-3-0)} and α -chloroal[de](#page-3-0)hydes,⁴ have been extremely successful. In 2007 Scheidt et al. reported the NHC-catalyzed generati[on](#page-3-0) [o](#page-3-0)f α , β -unsaturated acyl azolium from allylic alcohols under the oxidation of MnO₂.⁵ The generation of this $\alpha_1\beta$ unsaturated acyl azolium from α , β -unsaturated esters, 6 enals with oxidant,⁷ ynals,⁸ and α [-](#page-3-0)bromoenals⁹ has also been successfully established. As a versatile 1,3-biselectrop[hil](#page-3-0)e, the α,β-unsaturate[d](#page-3-0) acyl a[zo](#page-3-0)lium has been applie[d](#page-3-0) for the synthesis of various heterocycles via its $[3 + 2]^{10}$ and $[3 + 3]$ annulation reaction. 11

Cascade reactions, which form m[ult](#page-3-0)iple bonds in a single operatio[n,](#page-3-0) are very useful in organic synthesis,¹² especially for the synthesis of cyclic compounds.¹³ In 2011, Studer et al. reported the NHC-catalyzed cascade reaction [for](#page-3-0) the synthesis of indanodihydropyranones.¹⁴ In 2[013](#page-3-0), Romo et al. reported the chiral amine-catalyzed generation of α , β -unsaturated acyl ammonium from acyl chl[or](#page-3-0)ide and its following cascade Michael addition/aldol/lactonization for the synthesis of cyclopentane-fused β -lactones (Scheme 1, reaction a).¹⁵ At the same time, Lupton et al. reported the related NHCcatalyzed cascade of acyl fluoride with donor−ac[cep](#page-3-0)tor cyclopropanes (Scheme 1, reaction b).¹⁶ Recently, the cascade reaction of enals for bicyclic β-lactones was explored by Studer et al. (Scheme 1, reaction c).¹⁷ In t[his](#page-3-0) letter, we report the NHC-catalyzed cascade reaction for the synthesis of cyclopentane- an[d](#page-3-0) cyclohexane-fused δ -lactones (Scheme 1, reaction d), which is a key motif in many pharmaceutical compounds and natural products.¹⁸

Initially, the reaction of ε -oxo- γ , δ -malonate 1a and cinnamaldehyde 2a [wa](#page-3-0)s investigated as the model reaction (Table 1). We were encouraged to find that the desired bicyclic δ-lactone 4aa was obtained in 10% yield for the reaction [catalyzed](#page-1-0) by 10 mol % of achiral NHC A′ in the presence of 1.2

equiv of bisquinone 3 as the oxidant and 1,8-diazabicylo- [5.4.0]undec-7-ene (DBU) as a base (Table 1, entry 1). More encouraging, using the aminoindanol-derived tetracyclic NHC B¹⁹ as the catalyst led to better result[s, giving](#page-1-0) the product 4aa in 39% yield with exclusive diastereoselectivity but very low e[nan](#page-3-0)tioselectivity (entry 2).

Pioneered by Scheidt et al.,²⁰ Lewis acids were found to be efficient cooperative catalysts in the NHC-catalyzed reactions to improve the reactivity and s[ele](#page-3-0)ctivities. 21 Thus, several Lewis acids were then investigated for the reaction, and we were satisfied to find that the reaction with [50](#page-3-0) mol % LiCl as the additive gave the product in 73% yield with dramatically

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Table 1. Optimization of the Reaction Conditions^a

^aGeneral reaction conditions: 1a (0.2 mmol), 2a (1.2 equiv), cat A/B (10 mol %), base (50 mol %), LiCl (50 mol %), and bisquinone 3 (1.2 equiv). ^bIsolated yield. ^cDetermined by HPLC analysis. ^dNo LiCl was added.

improved enantioselectivity (93:7 er, entry 3). Screening of solvents revealed that the reaction succeeded in dichloromethane, toluene, and acetonitrile, while THF was the best choice (entries 3−6). Several bases were then examined for the reaction. It was found that all the bases DIPEA, DMAP, K_2CO_3 and Cs_2CO_3 worked well (entries 7–10), while the base $Cs₂CO₃$ afforded the best yield with a high enantiomeric ratio (entry 10). It should be noted that only a low yield and er resulted when LiCl was not added for the reaction with $Cs₂CO₃$ as the base (entry 11).

With suitable conditions in hand, the substrate scope of enals 2 was then investigated for the cascade reaction (Scheme 2). The aryl enals with electron-donating $(R = 4$ -MeOC₆H₄) or electron-withdrawing substituents ($R = 4$ -FC₆H₄, 4-ClC₆H₄, 4- BrC_6H_4 , 4-NO₂C₆H₄) all worked well, giving the desired bicyclic δ-enollactones 4ab−4af in high yields with excellent diastereo- and high enantioselectivities. The absolute configuration of bicyclic δ -lactone 4ae was established by the X-ray analysis of its crystal. In addition, the meta-substituent and ortho-substituents of enals were well tolerated (4ag−4aj). The enal with 2-furyl afforded the bicyclic δ -lactone 4ak in 90% yield with high enantioselectivity. Notably, enal with β -alkenyl worked well for the reaction albeit some excess of the enal (1.5 equiv) was required to give a good yield of product 4al. Furthermore, the reaction of enals with β -alkyl gave the products (4am−4an) in moderate yields with high diastereoand enantioslectivities when the reaction was carried out at 40 °C instead of room temperature.

Subsequently, variation of the malonates was examined (Scheme 3). As expected, both electron-donating and -withdrawing substituents of cinnamylmethyl malonates worked well to give the bicyclic δ-enollactones (4ba–4ea) in high yields with exclusive diastereo- and excellent enantioselectivities. εScheme 2. Variation of Enals

 a^a dr >20:1 unless otherwise specified. b^b 1.5 equiv of enal 2 and bisquinone 3 were used. Conducted at 40 °C.

Scheme 3. Variation of Malonates

Oxo-γ,δ-malonate with alkyl (R' = Me) also worked, affording the bicyclic δ-lactone 4fa in 87% yield with 92:8 er. Ethyl malonate worked as well as methyl malonate (4ga).

The reaction for the synthesis of cyclohexane-fused δ -lactone 6 from malonate 5 was then explored (Scheme 4). We were happy to find that the desired δ-lactone 6a was obtained in 68%

^aGeneral reactions: 5 (0.2 mmol), 2 (1.5 equiv), B (10 mol %), Cs₂CO₃ (50 mol %), LiCl (50 mol %), 3 (1.5 equiv). ^bReaction using 1.2 equiv of enal 2 and bisquinone 3. \degree Conducted at 40 \degree C.

yield with 97:3 er under the same conditions as in Schemes 2 and 3. The yield could be improved to 89% when excess enal 2a and bisquinone 3 (1.5 equiv) were applied. Various [enals were](#page-1-0) the[n t](#page-1-0)ested for the reaction. Enals with para-electron-donating $(R = 4$ -MeOC₆H₄) or electron-withdrawing substituents $(R =$ 4-ClC₆H₄, 4-BrC₆H₄) worked well, giving the bicyclic δ lactones (6b−6d) in high yields with high enantioselectivities. The enal with 2-furyl afforded the product 6e in 84% yield with 98:2 er. As expected, the reaction of enals with the β -alkenyl and alkyl chain gave the desired bicyclic δ -lactones 6f–6g in moderate yield with high enantioselectivities when carried out at 40 \degree C.

The plausible catalytic cycle is depicted in Figure 1. The addition of NHC to enal 2 gives the vinyl Breslow intermediate I, which is oxidized by bisquinone 3 to afford the key intermediate of α , β -unsaturated acyl azolium II. The reactive enolate III could be readily generated from malonate 1 or 5 in the presence of a base. 15 The Michael addition of enolate III to α,β-unsaturated acyl azolium II forms the first C−C bond and generates a new en[ola](#page-3-0)te IV. The intramolecular Michael addition of enolate IV makes the second C−C bond and furnishes the intermediate V with a cyclopentane or cyclohexane ring. The δ -lactonization of intermediate V finalizes the bicyclic δ-lactone 4 or 6 and regenerates the NHC catalyst. Currently, an alternative pathway with acylation followed by Clasien rearrangement, Michael addition, and lactonization could not be ruled out.

The possible transition states to rationalize the stereochemical outcome are d[ep](#page-3-0)icted in Figure 2. The coordination of the lithium with the enolate of malonate and α , β -unsaturated acyl azolium helps to assemble the complex and directs the enolate to attack α , β -unsaturated acyl azolium in a Michael addition manner from the less sterically demanded Re face (TS A). A chair-type conformation of intermediate IV may facilitate the second Michael addition and result in high stereoselectivity

Figure 1. Plausible catalytic cycle.

for the formation of the second C−C bond (TS B) to give the cyclopentane or cyclohexane ring.

In conclusion, the NHC-catalyzed cascade reaction via Michael/Michael addition/lactonization of enals with malonates to generate bicyclic δ -lactones was developed. The cyclopentane- and cyclohexane-fused δ -lactones with three continued stereocenters were obtained in high yields with excellent diastereo- and high enantioselectivities. In this reaction, two C−C bonds and one C−O bond were formed stereoselectively in one catalytic operation. Further development of the NHC-catalyzed cascade reactions is underway in our laboratory.

ASSOCIATED CONTENT

6 Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.5b02695.

Experimental details and NMR and HPLC spectra for obtained compounds (PDF) X-ray data for 4ae (CIF)

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Notes

The authors declare no competing financial interest.

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■ REFERENCES

(1) (a) DiRocco, D. A.; Rovis, T. Angew. Chem., Int. Ed. 2012, 51, 5904. (b) Ema, T.; Akihara, K.; Obayashi, R.; Sakai, T. Adv. Synth. Catal. 2012, 354, 3283. (c) Jia, M. Q.; You, S. L. ACS Catal. 2013, 3, 622. (d) Sun, L. H.; Liang, Z. Q.; Jia, W. Q.; Ye, S. Angew. Chem., Int. Ed. 2013, 52, 5803. (e) Thai, K.; Langdon, S. M.; Bilodeau, F.; Gravel, M. Org. Lett. 2013, 15, 2214. (f) Kang, B.; Sutou, T.; Wang, Y.; Kuwano, S.; Yamaoka, Y.; Takasu, K.; Yamada, K.-i. Adv. Synth. Catal. 2015, 357, 131. (g) Ramanjaneyulu, B. T.; Mahesh, S.; Vijaya Anand, R. Org. Lett. 2015, 17, 6.

(2) (a) DiRocco, D. A.; Rovis, T. J. Am. Chem. Soc. 2011, 133, 10402. (b) Fang, X. Q.; Chen, X. K.; Lv, H.; Chi, Y. R. Angew. Chem., Int. Ed. 2011, 50, 11782. (c) Kim, S. M.; Jin, M. Y.; Kim, M. J.; Cui, Y.; Kim, Y. S.; Zhang, L.; Song, C. E.; Ryu, D. H.; Yang, J. W. Org. Biomol. Chem. 2011, 9, 2069. (d) Patra, A.; Bhunia, A.; Biju, A. T. Org. Lett. 2014, 16, 4798. (e) Rafinski, Z.; Kozakiewicz, A.; Rafinska, K. ACS Catal. 2014, 4, 1404.

(3) (a) Burstein, C.; Glorius, F. Angew. Chem., Int. Ed. 2004, 43, 6205. (b) Sohn, S. S.; Rosen, E. L.; Bode, J. W. J. Am. Chem. Soc. 2004, 126, 14370. (c) Phillips, E. M.; Reynolds, T. E.; Scheidt, K. A. J. Am. Chem. Soc. 2008, 130, 2416. (d) Chan, A.; Scheidt, K. A. J. Am. Chem. Soc. 2008, 130, 2740. (e) Maji, B.; Ji, L.; Wang, S. M.; Vedachalam, S.; Ganguly, R.; Liu, X. W. Angew. Chem., Int. Ed. 2012, 51, 8276. (f) Zhao, Y. M.; Cheung, M. S.; Lin, Z. Y.; Sun, J. W. Angew. Chem., Int. Ed. 2012, 51, 10359. (g) Chen, X. Y.; Xia, F.; Cheng, J. T.; Ye, S. Angew. Chem., Int. Ed. 2013, 52, 10644. (h) Lv, H.; Jia, W. Q.; Sun, L. H.; Ye, S. Angew. Chem., Int. Ed. 2013, 52, 8607. (i) McCusker, E. O.; Scheidt, K. A. Angew. Chem., Int. Ed. 2013, 52, 13616. (j) Guo, C.; Schedler, M.; Daniliuc, C. G.; Glorius, F. Angew. Chem., Int. Ed. 2014, 53, 10232. (k) Wang, M.; Huang, Z. J.; Xu, J. F.; Chi, Y. R. J. Am. Chem. Soc. 2014, 136, 1214.

(4) (a) Reynolds, N. T.; Read de Alaniz, J. R.; Rovis, T. J. Am. Chem. Soc. 2004, 126, 9518. (b) He, M.; Uc, G. J.; Bode, J. W. J. Am. Chem. Soc. 2006, 128, 15088. (c) Vora, H. U.; Rovis, T. J. Am. Chem. Soc. 2010, 132, 2860. (d) Ni, Q. J.; Zhang, H.; Grossmann, A.; Loh, C. C. J.; Merkens, C.; Enders, D. Angew. Chem., Int. Ed. 2013, 52, 13562. (e) Yang, L.; Wang, F.; Lee, R.; Lv, Y.; Huang, K.-W.; Zhong, G. Org. Lett. 2014, 16, 3872. (f) Wang, D.-L.; Liang, Z.-Q.; Chen, K.-Q.; Sun, D.-Q.; Ye, S. J. Org. Chem. 2015, 80, 5900. (g) Dong, X.; Yang, W.; Hu, W.; Sun, J. Angew. Chem., Int. Ed. 2015, 54, 660.

(5) Maki, B. E.; Chan, A.; Phillips, E. M.; Scheidt, K. A. Org. Lett. 2007, 9, 371.

(6) (a) Ryan, S. J.; Candish, L.; Lupton, D. W. J. Am. Chem. Soc. 2009, 131, 14176. (b) Cheng, J.; Huang, Z.; Chi, Y. R. Angew. Chem., Int. Ed. 2013, 52, 8592.

(7) De Sarkar, S.; Studer, A. Angew. Chem., Int. Ed. 2010, 49, 9266.

(8) (a) Kaeobamrung, J.; Mahatthananchai, J.; Zheng, P. G.; Bode, J. W. J. Am. Chem. Soc. 2010, 132, 8810. (b) Zhu, Z. Q.; Xiao, J. C. Adv. Synth. Catal. 2010, 352, 2455.

(9) (a) Sun, F. G.; Sun, L. H.; Ye, S. Adv. Synth. Catal. 2011, 353, 3134. (b) Yao, C.; Wang, D.; Lu, J.; Li, T.; Jiao, W.; Yu, C. Chem. - Eur. J. 2012, 18, 1914.

(10) Chen, X. Y.; Gao, Z. H.; Song, C. Y.; Zhang, C. L.; Wang, Z. X.; Ye, S. Angew. Chem., Int. Ed. 2014, 53, 11611.

(11) (a) Chiang, P. C.; Rommel, M.; Bode, J. W. J. Am. Chem. Soc. 2009, 131, 8714. (b) Wanner, B.; Mahatthananchai, J.; Bode, J. W. Org. Lett. 2011, 13, 5378. (c) Kravina, A. G.; Mahatthananchai, J.; Bode, J. W. Angew. Chem., Int. Ed. 2012, 51, 9433. (d) Yetra, S. R.; Kaicharla, T.; Kunte, S. S.; Gonnade, R. G.; Biju, A. T. Org. Lett. 2013, 15, 5202. (e) Ni, Q. J.; Song, X. X.; Raabe, G.; Enders, D. Chem. - Asian J. 2014, 9, 1535. (f) Mao, J.-H.; Wang, Z.-T.; Wang, Z.-Y.; Cheng, Y. J. Org. Chem. 2015, 80, 6350. (g) Lu, Y. Y.; Tang, W. F.; Zhang, Y.; Du, D.; Lu, T. Adv. Synth. Catal. 2013, 355, 321.

(12) (a) Enders, D.; Grondal, C.; Hüttl, M. R. M. Angew. Chem., Int. Ed. 2007, 46, 1570. (b) Yu, X.; Wang, W. Org. Biomol. Chem. 2008, 6, 2037. (c) Xia, Y.; Zhang, Y.; Wang, J. ACS Catal. 2013, 3, 2586. (d) Volla, C. M. R.; Atodiresei, I.; Rueping, M. Chem. Rev. 2014, 114, 2390.

(13) (a) He, M.; Bode, J. W. J. Am. Chem. Soc. 2008, 130, 418. (b) Fang, X.; Jiang, K.; Xing, C.; Hao, L.; Chi, Y. R. Angew. Chem., Int. Ed. 2011, 50, 1910. (c) Ryan, S. J.; Candish, L.; Lupton, D. W. J. Am. Chem. Soc. 2011, 133, 4694. (d) Romanov-Michailidis, F.; Besnard, C.; Alexakis, A. Org. Lett. 2012, 14, 4906. (e) Zhang, H. R.; Dong, Z. W.; Yang, Y. J.; Wang, P. L.; Hui, X. P. Org. Lett. 2013, 15, 4750. (f) Zhou, B.; Luo, Z.; Li, Y. C. Chem. - Eur. J. 2013, 19, 4428. (g) Candish, L.; Levens, A.; Lupton, D. W. J. Am. Chem. Soc. 2014, 136, 14397. (h) Yang, Y. J.; Zhang, H. R.; Zhu, S. Y.; Zhu, P.; Hui, X. P. Org. Lett. 2014, 16, 5048. (i) Zhao, Y.; Wang, Z. T.; Cheng, Y. Adv. Synth. Catal. 2014, 356, 2580.

(14) Biswas, A.; Sarkar, S. D.; Frö hlich, R.; Studer, A. Org. Lett. 2011, 13, 4966.

(15) Liu, G.; Shirley, M. E.; Van, K. N.; McFarlin, R. L.; Romo, D. Nat. Chem. 2013, 5, 1049.

(16) Candish, L.; Lupton, D. W. J. Am. Chem. Soc. 2013, 135, 58.

(17) Bera, S.; Samanta, R. C.; Daniliuc, C. G.; Studer, A. Angew. Chem., Int. Ed. 2014, 53, 9622.

(18) (a) McElvain, S. M.; Eisenbraun, E. J. J. Am. Chem. Soc. 1955, 77, 1599. (b) Willot, M.; Radtke, L.; Kö nning, D.; Frö hlich, R.; Gessner, V. H.; Strohmann, C.; Christmann, M. Angew. Chem., Int. Ed. 2009, 48, 9105. (c) Candish, L.; Lupton, D. W. Org. Lett. 2010, 12, 4836. (d) Zheng, C.; Dubovyk, I.; Lazarski, K. E.; Thomson, R. J. J. Am. Chem. Soc. 2014, 136, 17750.

(19) (a) Kerr, M. S.; Read de Alaniz, J.; Rovis, T. J. Am. Chem. Soc. 2002, 124, 10298. (b) He, M.; Struble, J. R.; Bode, J. W. J. Am. Chem. Soc. 2006, 128, 8418.

(20) (a) Cardinal-David, B.; Raup, D. E. A.; Scheidt, K. A. J. Am. Chem. Soc. 2010, 132, 5345. (b) Raup, D. E. A.; Cardinal-David, B.; Holte, D.; Scheidt, K. A. Nat. Chem. 2010, 2, 766. (c) Dugal-Tessier, J.; O'Bryan, E. A.; Schroeder, T. B. H.; Cohen, D. T.; Scheidt, K. A. Angew. Chem., Int. Ed. 2012, 51, 4963. (d) Lee, A.; Scheidt, K. A. Angew. Chem., Int. Ed. 2014, 53, 7594.

(21) (a) Rong, Z. Q.; Jia, M. Q.; You, S. L. Org. Lett. 2011, 13, 4080. (b) Mo, J.; Chen, X.; Chi, Y. R. J. Am. Chem. Soc. 2012, 134, 8810. (c) Mo, J.; Shen, L.; Chi, Y. R. Angew. Chem., Int. Ed. 2013, 52, 8588. (d) Xiao, Z. X.; Yu, C. X.; Li, T. J.; Wang, X. S.; Yao, C. S. Org. Lett. 2014, 16, 3632. (e) Cheng, J.-T.; Chen, X.-Y.; Gao, Z.-H.; Ye, S. Eur. J. Org. Chem. 2015, 2015, 1047. (f) Wu, Z.; Li, F.; Wang, J. Angew. Chem., Int. Ed. 2015, 54, 1629.

(22) (a) See ref 8a. (b) Li, G.-T.; Gu, Q.; You, S.-L. Chem. Sci. 2015, 6, 4273.